

## Investigation of a New Family of Chiral Ligands for Enantioselective Catalysis via Parallel Synthesis and High-Throughput Screening

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Applications of combinatorial chemistry are widespread and cover fields as diverse as drug discovery and optimization,<sup>1</sup> material science,<sup>2</sup> studies of molecular recognition,<sup>3</sup> and the development of new catalysts.<sup>4,5</sup> In particular, the possibility of high-throughput catalyst screening for the development and optimization of enantioselective reactions has generated a lot of excitement.<sup>5d–j</sup> Two different basic approaches have been considered: optimization of the reaction conditions (solvents, temperatures, stoichiometries, different ligands, or metal ions)<sup>5e</sup> and the synthesis of new ligands via a modular building block strategy in which the stereoelectronic properties of a metal binding site (e.g., a diphosphine<sup>5f</sup> or a Schiff-base<sup>5h–j</sup>) are tuned by variation of the substituents and side chains. In the case of screening members of a library containing ligands for enantioselective catalysis, the identification of a hit requires a demanding selection procedure, since the screen is ultimately catalysis of a reaction and analysis of its stereochemical outcome. For this reason, a combinatorial system is usually chosen that allows the synthesis of discrete isolated compounds. Parallel

synthesis (as opposed to the “split and pool” methodology)<sup>5c</sup> allows one to know the identity of each ligand and keeps the ligands separate so that screening of individual ligand metal complexes can be performed.

We have developed a new family of chiral ligands based on a modular building block strategy and on the use of a disulfonamide as metal chelating unit, for which a number of examples are already known.<sup>6</sup> These ligands are synthesized by coupling commercially available vicinal diamines **1** and a novel class of chiral N-protected  $\beta$ -amino sulfonyl chlorides of general formula **2** (Scheme 1), obtained in high yields from L- $\alpha$ -amino acids via a straightforward synthetic protocol.<sup>7</sup> For the construction of the library (30 compounds), we used the sulfonyl chlorides derived from L-alanine (**2g**), L-valine (**2h**), L-leucine (**2i**), L-phenylalanine (**2j**), and L-proline (**2k**).<sup>8</sup> As for the diamine part of the library, we employed two vicinal diamine scaffolds: 1,2-diaminocyclohexane<sup>9</sup> (**1a–d**) and 1,2-diphenylethylenediamine<sup>10</sup> (**1e–f**), for which effective use in the fields of asymmetric synthesis and molecular recognition is well documented. In the case of 1,2-diaminocyclohexane, besides the chiral *trans-R,R* and *trans-S,S*, we have used the achiral *cis-R,S* and the racemic ( $\pm$ ) *trans* structures to take into consideration all the possible stereochemical combinations, including those that might seem odd at first glance, and to take advantage of possible cooperative effects arising from the formation of aggregates.<sup>11</sup>

In principle, the synthesis and subsequent test of the library could be accomplished with the ligands bound to a solid support. However, the need for an additional handle to attach the diamine scaffold to the support and the controversial role of the solid matrix on the yields and enantiomeric ratios (er's) of the catalyzed reactions<sup>5h,12</sup> make this route less attractive. On the other hand, the classical synthesis in solution suffers from a major disadvantage, i.e., the necessity of workup and purification (chromatography), which makes parallel chemistry impractical. An answer to these problems was found with the use of solid-phase extraction (SPE) techniques.<sup>13</sup> For the coupling reactions (Scheme 1), we treated the vicinal diamines **1** with an excess of sulfonyl chlorides **2** to ensure complete conversion and

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(1) (a) Gallop, M. A.; Barrett R. W.; Dower, W. J.; Fodor, S. P. A.; Gordon, E. M. *J. Med. Chem.* **1994**, *37*, 1233. (b) Gordon, E. M.; Barrett R. W.; Dower, W. J.; Fodor, S. P. A.; Gallop, M. A. *J. Med. Chem.* **1994**, *37*, 1385. (c) Terrett, N. K.; Gardner, M.; Gordon, D. W.; Kobylecki, R. J.; Steele, J. *Tetrahedron* **1995**, *51*, 8135. (d) Balkenhol, F.; von dem Bussche-Hünnefeld, C.; Lansky, A.; Zechel, C. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 2288. (e) Thompson, L. A.; Ellman, J. A. *Chem. Rev.* **1996**, *96*, 555. (f) Lam, K. S.; Lebl, M.; Krčňák, V. *Chem. Rev.* **1997**, *97*, 411.

(2) (a) Xiang, X.-D.; Sun, X.; Briceño, G.; Lou, Y.; Wang, K.-A.; Chang, H.; Wallace-Freedman, W. G.; Chen, S.-W.; Schultz, P. G. *Science* **1995**, *268*, 1738. (b) Danielson, E.; Devenney, M.; Giaquinta, D. M.; Golden, J. H.; Haushalter, R. C.; McFarland, E. W.; Poojary, D. M.; Reaves, C. M.; Weinberg, W. H.; Wu, X. D. *Science* **1998**, *279*, 837.

(3) Still, W. C. *Acc. Chem. Res.* **1996**, *29*, 155.

(4) For review articles, see: (a) Gennari, C.; Nestler, H. P.; Piarulli, U.; Salom, B. *Liebigs Ann./Recueil* **1997**, *637*. (b) Stinson, S. C. *Chem. Eng. News* **1996**, *74* (April 15), 24. (c) Borman, S. *Chem. Eng. News* **1996**, *74* (Nov 4), 37. (d) Wilson, E. K. *Chem. Eng. News* **1997**, *75* (Dec 8), 24. (e) Brady, P. A.; Sanders, J. K. M. *Chem. Soc. Rev.* **1997**, *26*, 327.

(5) For research papers, see: (a) Menger, F. M.; Eliseev, A. V.; Migulin, V. A. *J. Org. Chem.* **1995**, *60*, 6666. (b) Menger, F. M.; West, C. A.; Ding, J. *Chem. Commun.* **1997**, 633. (c) Taylor, S. J.; Morken, J. P. *Science* **1998**, *280*, 267. (d) Liu, G.; Ellman, J. A. *J. Org. Chem.* **1995**, *60*, 7712. (e) Burgess, K.; Lim, H.; Porte, G.; Sulikowski, G. A. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 220. (f) Gilbertson, S. R.; Wang, X. *Tetrahedron Lett.* **1996**, *37*, 6475. (g) Gilbertson, S. R.; Chang, C. T. *Chem. Commun.* **1997**, 975. (h) Cole, B. M.; Shimizu, K. D.; Krueger, C. A.; Harrity, J. P. A.; Snapper, M. L.; Hoveyda, A. H. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 1668. (i) Shimizu, K. D.; Cole, B. M.; Krueger, C. A.; Kuntz, K. W.; Snapper, M. L.; Hoveyda, A. H. *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 1704. (j) Sigman, M. S.; Jacobsen, E. N. *J. Am. Chem. Soc.* **1998**, *120*, 4901.

(6) For Et<sub>2</sub>Zn additions to aldehydes, see: (a) Takahashi, H.; Kawakita, T.; Yoshioka, M.; Kobayashi, S.; Ohno, M. *Tetrahedron Lett.* **1989**, *30*, 7095. (b) Takahashi, H.; Kawakita, T.; Ohno, M.; Yoshioka, M.; Kobayashi, S. *Tetrahedron* **1992**, *48*, 5691. (c) Zhang, X.; Guo, C. *Tetrahedron Lett.* **1995**, *36*, 4947. (d) Lutz, C.; Knochel, P. *J. Org. Chem.* **1997**, *62*, 7895. (e) Ramón, D. J.; Yus, M. *Tetrahedron: Asymmetry* **1997**, *8*, 2479. (f) Cernerud, M.; Skrinning, A.; Bergère, I.; Moberg, C. *Tetrahedron: Asymmetry* **1997**, *8*, 3437. (g) Halm, C.; Kurth, M. J. *Angew. Chem., Int. Ed. Engl.* **1998**, *37*, 510. For cyclopropanation reactions, see: (h) Takahashi, H.; Yoshioka, M.; Shibasaki, M.; Ohno, M.; Imai, N.; Kobayashi, S. *Tetrahedron* **1995**, *51*, 12013 and references therein. (i) Denmark, S. E.; O'Connor, S. P. *J. Org. Chem.* **1997**, *62*, 584. (j) Denmark, S. E.; O'Connor, S. P. *J. Org. Chem.* **1997**, *62*, 3390 and references therein. (k) Imai, N.; Sakamoto, K.; Maeda, M.; Kouge, K.; Yoshizane, K.; Nokami, J. *Tetrahedron Lett.* **1997**, *38*, 1423.

(7) (a) Gude, M.; Piarulli, U.; Potenza, D.; Salom, B.; Gennari, C. *Tetrahedron Lett.* **1996**, *37*, 8589. (b) Gude, M.; Piarulli, U.; Potenza, D.; Gennari, C. *Chem. Eur. J.* **1998**, in press.

(8) In the case of proline, problems with the preparation of the *N*-Boc-protected sulfonyl chloride via the usual route suggested the alternative use of *N*-Cbz protection.

(9) Bennani, Y. L.; Hanessian, S. *Chem. Rev.* **1997**, *97*, 3161.

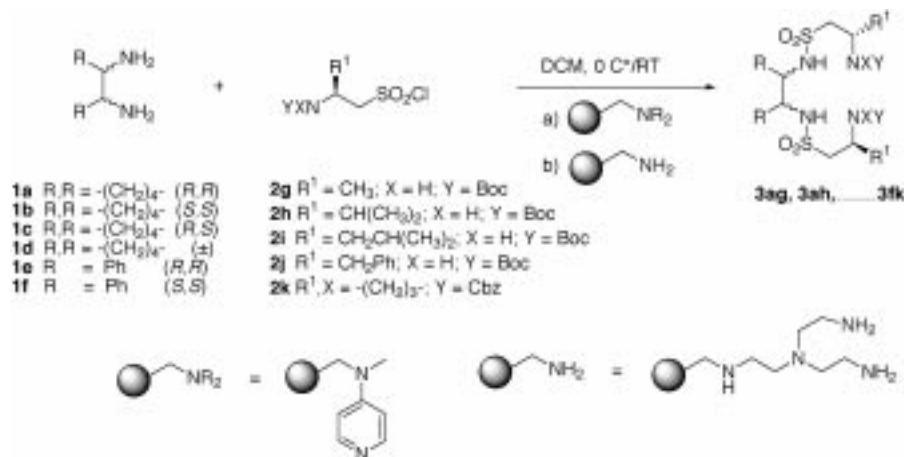
(10) (a) Mangeney, P.; Tejero, T.; Alexakis, A.; Grosjean, F.; Normant, J. *Synthesis* **1988**, 255. (b) Pikul, S.; Corey, E. J. *Org. Synth.* **1993**, *71*, 22.

(11) (a) Mikami, K.; Matsukawa, S.; Volk, T.; Terada, M. *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 2768. (b) Ohkuma, T.; Doucet, H.; Pham, T.; Mikami, K.; Korenaga, T.; Tearda, M.; Noyori, R. *J. Am. Chem. Soc.* **1998**, *120*, 1086.

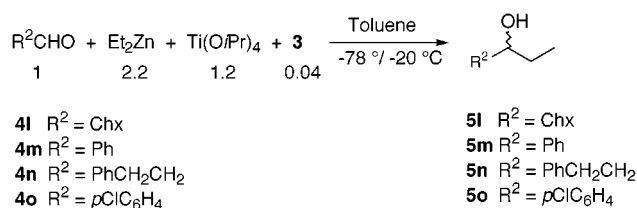
(12) Hodge, P. *Chem. Soc. Rev.* **1997**, *26*, 417.

(13) (a) Lawrence M. R.; Biller, S. A.; Fryszman, O. M.; Poss, M. A. *Synthesis* **1997**, 553. (b) Flynn, D. L.; Crich, J. Z.; Devraj, R. V.; Hockerman, S. L.; Parlow, J. J.; South, M. S.; Woodward, S. *J. Am. Chem. Soc.* **1997**, *119*, 4874. (c) Parlow, J. J.; Mischke, D. A.; Woodward, S. S. *J. Org. Chem.* **1997**, *62*, 5908. (d) Booth, R. J.; Hodges, J. C. *J. Am. Chem. Soc.* **1997**, *119*, 4882.

Scheme 1



Scheme 2



used SPE methodology to avoid purification of the sulfonamide products **3**. The reaction was run in DCM in the presence of polymer bound “dimethylamino pyridine”<sup>13</sup> to catalyze the coupling reaction and scavenge the liberated HCl; after all the diamine had been converted into the disulfonamide derivative, the excess of sulfonyl chloride was removed by reaction with solid-phase bound tris(2-aminoethyl)amine.<sup>13b</sup>

As a first test for the disulfonamide library, we have chosen the Ti(O-*i*-Pr)<sub>4</sub>-mediated addition of Et<sub>2</sub>Zn to aldehydes. In the enantioselective version of this reaction, catalytic amounts of chiral disulfonamides have given good to excellent er's,<sup>6a–g</sup> and a fine-tuning of the ligand structure was shown to be necessary for achieving good enantioselectivities starting from different aldehydes. The tests were run in a parallel format (30 different reaction vessels, Scheme 2) on a 0.1 mmol scale with 4 μmol of the chiral ligand on a mixture of four different aldehydes, and the er's were determined by analysis of the crude reaction mixtures using a chiral capillary GC column (120 different results).<sup>14</sup>

Following a different selection approach for ligand optimization, we used a mixture of the five ligands derived from every single diamine (e.g., **3ag–3ak**) with a single aldehyde (six different reaction vessels) for the evaluation of the influence of the scaffold on enantioselectivity.

The screening revealed a number of interesting and somewhat unexpected results: (1) The best ligand for this reaction is **3bj** (i.e., 1*S*,2*S*-diaminocyclohexane as diamine

scaffold and the sulfonyl chloride derived from L-phenylalanine). With **3bj**, excellent enantiomeric ratios were obtained with both aromatic and aliphatic aldehydes in favor of the (*R*)-alcohol [*R*:*S* = 93:7 (**5l**), 98:2 (**5m**), 98:2 (**5n**), 97:3 (**5o**)]. Ligand **3bj** was then purified (chromatography) and fully characterized, and the screening results were confirmed by reaction with the four separate aldehydes on a preparative scale (1 mmol). (2) The influence of the different β-aminosulfonyl side chains in controlling the er's is as follows: R<sup>1</sup> = CH<sub>2</sub>Ph (**2j**) > CH<sub>3</sub> (**2g**) ≥ *i*-Bu (**2i**) > *i*-Pr (**2h**) ≫ (CH<sub>2</sub>)<sub>3</sub> (**2k**). (3) The influence of the different scaffolds in controlling the er's is as follows: *trans*-(1*S*,2*S*)-diaminocyclohexane (**1b**) > *cis*-diaminocyclohexane (**1c**) ≈ (±)-racemic-1,2-diaminocyclohexane (**1d**) > (1*R*,2*R*)-diphenylethylenediamine (**1e**) ≈ (1*S*,2*S*)-diphenylethylenediamine (**1f**) > *trans*-(1*R*,2*R*)-diaminocyclohexane (**1a**). (4) With the *cis* and the racemic scaffolds, moderate enantiomeric ratios were obtained in favor of the (*R*)-alcohol [**3ci**, *R*:*S* = 87:13 (**5m**), 88:12 (**5o**); **3dj**, *R*:*S* = 80:20 (**5m**), 81:19 (**5n**); **3di**, *R*:*S* = 80:20 (**5o**)]. (5) With the (*R,R*)-diphenylethylenediamine scaffold, one single reasonably high enantiomeric ratio was obtained [**3ej** *R*:*S* = 89:11 (**5o**)].

The use of the chiral ligand library (**3ag–3fk**) in other enantioselective reactions of synthetic importance is presently under investigation.

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**Supporting Information Available:** Experimental procedures for the library synthesis (30 compounds **3ag–3fk**) and screening and additional experimental procedures including physical and spectroscopic data for compounds **2g–2k**, **3bj**, and **5l–5o** (8 pages).

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(14) The GC conditions were optimized in order to have baseline separation of the racemic mixtures of the four different alcohols (eight separate peaks).